

Data collection and cell refinement: *MSC/AFC Diffractometer Control Software* (Molecular Structure Corporation, 1988). Data reduction: *TEXSAN PROCESS* (Molecular Structure Corporation, 1985). Program(s) used to solve structure: *SHELXS86* (Sheldrick, 1985). Program(s) used to refine structure: *TEXSAN LS*. Molecular graphics: *TEXSAN*, *ORTEP* (Johnson, 1965). Software used to prepare material for publication: *TEXSAN FINISH*.

We gratefully acknowledge financial support of our program from the Alfred P. Sloan Foundation and American Heart Association, Kansas Affiliate.

Lists of structure factors, anisotropic displacement parameters and H-atom coordinates have been deposited with the IUCr (Reference: CR1101). Copies may be obtained through The Managing Editor, International Union of Crystallography, 5 Abbey Square, Chester CH1 2HU, England.

References

- Antoun, S. (1987). *Collect. Czech. Chem. Commun.* **52**, 162–165.
 Dixon, E. A., Fischer, A. & Robinson, F. P. (1981). *Can. J. Chem.* **59**, 2629–2641.
 Dougherty, D. A., Choi, C. S., Kaupp, G., Buda, A. B., Rudzinski, J. M. & Osawa, E. (1986). *J. Chem. Soc. Perkin Trans. 2*, pp. 1063–1070.
 Johnson, C. K. (1965). *ORTEP*. Report ORNL-3794. Oak Ridge National Laboratory, Tennessee, USA.
 Kim, H., Waldow, D. A., Han, C. C., Qui, T.-C. & Yamamoto, M. (1991). *Polym. Commun.* **32**, 108–112.
 Miki, S., Asako, Y. & Yoshida, Z.-I. (1987). *Chem. Lett.* pp. 195–198, and references 1–3 therein.
 Molecular Structure Corporation (1985). *TEXSAN. TEXRAY Structure Analysis Package*. MSC, 3200 Research Forest Drive, The Woodlands, TX 77381, USA.
 Molecular Structure Corporation (1988). *MSC/AFC Diffractometer Control Software*. MSC, 3200 Research Forest Drive, The Woodlands, TX 77381, USA.
 Nishimura, J., Takeuchi, M., Takahashi, H. & Sato, M. (1990). *Tetrahedron Lett.* **31**, 2911–2914.
 Sheldrick, G. M. (1985). *SHELXS86. Crystallographic Computing 3*, edited by G. M. Sheldrick, C. Krüger & R. Goddard, pp. 175–189. Oxford Univ. Press.
 Shinmyozu, T., Inazu, T. & Yoshino, T. (1978). *Chem. Lett.* pp. 405–408, and references therein.
 Tazuke, S. & Watanabe, H. (1982). *Tetrahedron Lett.* **23**, 197–200, and references 1–3 therein.
 Vollhardt, K. P. C. (1987). *Organic Chemistry*, p. 121. New York: W. H. Freeman.

Acta Cryst. (1994). **C50**, 1093–1095

(R)-(+)-2-Phenylpropyl *p*-Toluenesulfonate, C₁₆H₁₈O₃S

HUGO GALLARDO AND ALOIR A. MERLO

Depto. Química – UFSC, 88040-900, Florianópolis, SC, Brazil

IVO VENCATO

Depto. Física – UFSC, 88040-900, Florianópolis, SC, Brazil

YVONNE P. MASCARENHAS

Instituto de Física e Química de São Carlos – USP, 13560-970, São Carlos, SP, Brazil

(Received 27 August 1993; accepted 24 December 1993)

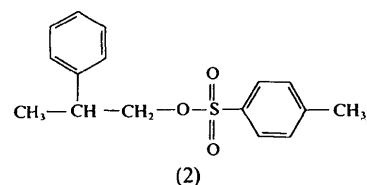
Abstract

The structure of an important precursor of a ferroelectric liquid crystal is reported. The two phenyl rings are nearly coplanar, having a dihedral angle of 5.0°, and are at a distance of 3.79 (2) Å from each other.

Comment

In connection with a project on the design and synthesis of ferroelectric liquid crystals (FLC), we required an efficient route to obtain several chiral precursors (Merlo, Gallardo, Taylor & Kroin, 1993). (*S*)-(–)-Ethyl lactate is particularly attractive as a chiral synthon as it is readily available in both enantiomeric forms with very high optical purity and, therefore, has been used in numerous applications (Seuring & Seebach, 1977). Recently, we described an efficient method for the preparation of (*R*)-(–)-*p*-(2-methylhexyl)aniline, (1), from (*S*)-(–)-ethyl lactate (Merlo & Gallardo, 1993), which is of considerable interest as a building block for liquid-crystalline materials having ferroelectric properties.

We have also found that the optical purity of the aniline enantiomeric to (1) can be ensured by using (*R*)-(+)-2-phenylpropyl *p*-toluenesulfonate (2) as a crystalline intermediate (Merlo & Gallardo, 1993). The molecular structure determination of the title compound (2) was undertaken by X-ray diffraction. The stereochemistry of



the title compound and its atomic numbering scheme are illustrated in Fig. 1, which shows the *cis* relationship between the two aromatic rings. The absolute configuration was assigned to agree with the known chirality at C(2) arising from its precursor (*R*)-(-)-*p*-(2-methylhexyl)aniline.

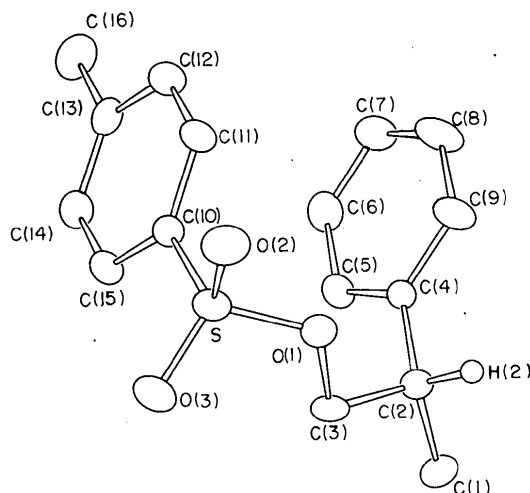


Fig. 1. Perspective view of the molecule with atomic numbering. The atom H(2) is represented by a sphere of arbitrary radius.

Experimental

(*R*)-(-)-2-Phenylpropanol (55 mmol, 7.5 g) was dissolved in dry pyridine (50 ml) and *p*-toluenesulfonyl chloride (60 mmol, 11.40 g) was added in small portions at 273 K. After 2 h at 273 K the mixture was poured onto ice and water, then extracted with chloroform. The organic solution was washed once with hydrochloric acid 10% (100 ml), twice with water (100 ml) and once with brine. After drying (sodium sulfate) and evaporation of the solvent, the product was recrystallized three times from hexane to yield (64%) (*R*)-(+)-2-phenylpropyl *p*-toluenesulfonate; $[\alpha]_D^{25} = +9.0$ (c 0.2, CHCl₃), m.p. 335 K. ¹H NMR (60 MHz, TMS, CDCl₃): δ = 1.2 (*d*, 3H, *J* = 7 Hz, CH₃); 3.0 (*m*, 1H, CH); 4.1 (*d*, 2H, *J* = 7 Hz, CH₂); 7.3 (*s*, 5H, Ar); 7.4 (*d*, 2H, *J* = 8 Hz, Ar); 7.8 (*d*, 2H, *J* = 8 Hz, Ar). Analysis: calculated for C₁₆H₁₈SO₃, C 66.20, H 6.20, S 11.03%; found, C 66.23, H 6.23, S 11.07%.

Crystal data

C₁₆H₁₈O₃S
M_r = 290.4
 Monoclinic
*P*2₁
a = 5.923 (1) Å
b = 15.708 (2) Å
c = 8.421 (2) Å
 β = 105.70 (2)°
V = 754.2 (3) Å³
Z = 2
D_x = 1.28 Mg m⁻³

Mo K α radiation
 λ = 0.7107 Å
 Cell parameters from 25 reflections
 θ = 5.7–15.5°
 μ = 0.176 mm⁻¹
T = 296 K
 Irregular
 0.25 × 0.18 × 0.35 mm
 Colourless

Data collection

Enraf-Nonius CAD-4 diffractometer
 $\omega/2\theta$ scans
 Absorption correction: none
 4400 measured reflections
 3279 independent reflections
 1977 observed reflections
 $[F > 6.0\sigma(F)]$

$R_{\text{int}} = 0.022$
 $\theta_{\text{max}} = 29.15^\circ$
 $h = -9 \rightarrow 11$
 $k = -1 \rightarrow 25$
 $l = 0 \rightarrow 15$
 2 standard reflections
 frequency: 30 min
 intensity variation: 1.5%

Refinement

Refinement on *F*
R = 0.055
 $wR = 0.065$
S = 3.84
 1977 reflections
 181 parameters
 H-atom parameters not refined
 $w = 3.012/[\sigma^2(F) + 0.0051F^2]$

$(\Delta/\sigma)_{\text{max}} = 0.30$
 $\Delta\rho_{\text{max}} = 0.34 \text{ e \AA}^{-3}$
 $\Delta\rho_{\text{min}} = -0.23 \text{ e \AA}^{-3}$
 Extinction correction: none
 Atomic scattering factors from *International Tables for X-ray Crystallography* (1974, Vol. IV)

Table 1. Fractional atomic coordinates and equivalent isotropic displacement parameters (Å²)

$$B_{\text{eq}} = (8\pi^2/3)\sum_i \sum_j U_{ij} a_i^* a_j^* a_i \cdot a_j$$

	<i>x</i>	<i>y</i>	<i>z</i>	<i>B_{eq}</i>
S	0.2960 (2)	0.2625	0.4280 (1)	4.20 (4)
O(1)	0.3536 (6)	0.1652 (3)	0.4506 (4)	4.4 (1)
O(2)	0.0543 (5)	0.2676 (4)	0.4233 (5)	5.7 (2)
O(3)	0.3743 (7)	0.2946 (3)	0.2937 (4)	5.6 (2)
C(1)	0.855 (1)	0.0153 (4)	0.4783 (8)	5.5 (2)
C(2)	0.6534 (9)	0.0584 (3)	0.5320 (6)	4.1 (2)
C(3)	0.573 (1)	0.1349 (4)	0.4220 (5)	4.4 (2)
C(4)	0.7197 (9)	0.0830 (3)	0.7111 (5)	3.8 (2)
C(5)	0.911 (1)	0.1332 (4)	0.7786 (7)	4.9 (2)
C(6)	0.973 (1)	0.1560 (4)	0.9436 (7)	5.7 (2)
C(7)	0.836 (1)	0.1278 (5)	1.0425 (7)	6.2 (3)
C(8)	0.651 (1)	0.0773 (6)	0.9778 (7)	7.3 (4)
C(9)	0.587 (1)	0.0543 (5)	0.8136 (7)	5.4 (2)
C(10)	0.4639 (7)	0.3076 (3)	0.6123 (5)	3.5 (1)
C(11)	0.386 (1)	0.2997 (4)	0.7514 (7)	4.9 (2)
C(12)	0.517 (1)	0.3329 (4)	0.8994 (7)	5.2 (2)
C(13)	0.724 (1)	0.3750 (3)	0.9076 (7)	4.6 (2)
C(14)	0.7974 (9)	0.3842 (4)	0.7656 (7)	4.7 (2)
C(15)	0.6685 (8)	0.3503 (4)	0.6180 (7)	4.4 (2)
C(16)	0.872 (1)	0.4102 (5)	1.0708 (9)	6.4 (3)

Table 2. Selected geometric parameters (Å, °)

O(1)—S	1.566 (4)	C(3)—O(1)	1.464 (6)
O(2)—S	1.423 (3)	C(2)—C(1)	1.543 (8)
O(3)—S	1.426 (4)	C(3)—C(2)	1.514 (8)
C(10)—S	1.751 (4)	C(4)—C(2)	1.503 (6)
O(2)—S—O(1)	104.0 (3)	C(3)—O(1)—S	118.1 (3)
O(3)—S—O(1)	109.6 (2)	C(3)—C(2)—C(1)	108.0 (4)
O(3)—S—O(2)	119.9 (3)	C(4)—C(2)—C(1)	113.2 (5)
C(10)—S—O(1)	103.5 (2)	C(4)—C(2)—C(3)	111.3 (4)
C(10)—S—O(2)	109.1 (2)	C(2)—C(3)—O(1)	107.7 (4)
C(10)—S—O(3)	109.4 (2)		

The *y* coordinate of S was held to fix the origin during refinement. H atoms were included at geometrically idealized positions (C—H 0.98 Å, *U* = 0.0800 Å²); their parameters were not refined, but their coordinates were recalculated af-

ter each cycle. Most of the calculations were performed on an IBM 3090 computer at the Universidade Federal de Santa Catarina. Data collection, cell refinement and data reduction: *SDP-Plus* (Frenz, 1985). Program(s) used to solve structure: *SHELX86* (Sheldrick, 1985). Program(s) used to refine structure: *SHELX76* (Sheldrick, 1976). Molecular graphics: *ORTEPII* (Johnson, 1976).

This work was supported by grants from the CNPq, the FAPESP and the FUNCITEC.

Lists of structure factors, anisotropic displacement parameters and H-atom coordinates have been deposited with the IUCr (Reference: HA1081). Copies may be obtained through The Managing Editor, International Union of Crystallography, 5 Abbey Square, Chester CH1 2HU, England.

References

- Frenz, B. A. (1985). *Enraf-Nonius SDP-Plus Structure Determination Package*. Enraf-Nonius, Delft, The Netherlands.
 Johnson, C. K. (1976). *ORTEPII*. Report ORNL-5138. Oak Ridge National Laboratory, Tennessee, USA.
 Merlo, A. A. & Gallardo, H. (1993). *Synth. Commun.* **15**, 2159–2169.
 Merlo, A. A., Gallardo, H., Taylor, T. R. & Kroin, T. (1993). *Mol. Cryst. Liq. Cryst.* In the press.
 Seuring, B. & Seebach, D. (1977). *Helv. Chim. Acta*, **60**, 1175.
 Sheldrick, G. M. (1976). *SHELX76. Program for Crystal Structure Determination*. Univ. of Cambridge, England.
 Sheldrick, G. M. (1985). *SHELXS86. Program for the Solution of Crystal Structures*. Univ. of Göttingen, Germany.

Acta Cryst. (1994). **C50**, 1095–1097

Methyl *N,N*-Diallyl-*O*-*tert*-butyltyrosyl- α -aminoisobutyryl- α -aminoisobutyrate, $C_{28}H_{43}N_3O_5$

JUDITH L. FLIPPEN-ANDERSON, CLIFFORD GEORGE AND JEFFREY R. DESCHAMPS

Laboratory for the Structure of Matter, Naval Research Laboratory, Washington, DC 20375, USA

(Received 3 August 1993; accepted 4 November 1993)

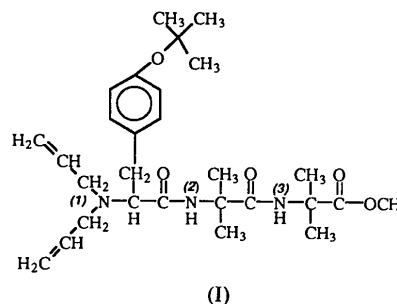
Abstract

In this *N*- and *C*-protected tripeptide the nine backbone atoms are essentially planar and fully extended except for *N1* and *C3'* which lie below the plane and are folded in towards it. The tyrosine side chain extends away from the opposite side of the plane. There is one $(N-H)_i \cdots (O=C)_i$ intramolecular hydrogen bond and a weak $(N-H)_{i+2} \cdots (O=C)_i$ intermolecular hydrogen bond [where $(i) = 2$].

©1994 International Union of Crystallography
 Printed in Great Britain – all rights reserved

Comment

N,N-Diallylenkephalin analogs have been shown to have activity as δ -selective opiate antagonists (Thorner *et al.*, 1986). The title compound, (I), was synthesized at the Research Triangle Institute, North Carolina, as part of a program designed to study the relative antagonist activity within this family of compounds.



The shape of the peptide backbone is defined by the following torsion angles: $\psi_1 = 20.6$ (6), $\omega_2 = -172.2$ (5), $\varphi_2 = 179.2$ (5), $\psi_2 = 172.2$ (5), $\omega_3 = 176.2$ (4) and $\varphi_3 = -54.7$ (6) $^\circ$. The tyrosyl side chain extends away from the main peptide backbone, $\chi_1 = -144.1^\circ$, and is rotated about the $C1\beta-C1\gamma$ bond such that the aromatic ring is approximately *gauche* to the plane formed by backbone atoms $C1\alpha-C3\alpha$ with the angle between the two planes being 41.5° . The hydrogen-bond parameters, $N \cdots O$, $H \cdots O$ and $N-H \cdots O$ are 2.566, 2.091 Å and 113.6 $^\circ$, respectively, for the intramolecular $N2 \cdots O2$ hydrogen bond, and 3.242, 2.531 Å and 139.5 $^\circ$, respectively, for the weak intermolecular $N3 \cdots O1$ hydrogen bond. There are no other intermolecular approaches less than van der Waals separations. The *tert*-butyl group on $O1\zeta$ is disordered with a ratio of 4:1 for the high to low occupancy.

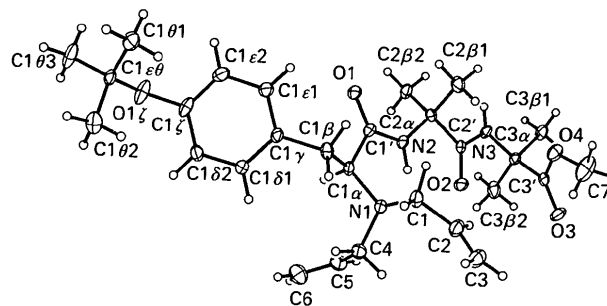


Fig. 1. Structure of the tripeptide showing 20% probability displacement ellipsoids. The disordered *tert*-butyl group on $C1\epsilon\theta$ is not shown.

Experimental

Crystal data

$C_{28}H_{43}N_3O_5$
 $M_r = 501.65$

Cu $K\alpha$ radiation
 $\lambda = 1.54178 \text{ \AA}$

Acta Crystallographica Section C
 ISSN 0108-2701 ©1994